

POTENTIAL CHEMICAL IMPACTS OF CO₂ LEAKAGE ON UNDERGROUND SOURCE OF DRINKING WATER (USDWs) ASSESSED BY QUANTITATIVE RISK ANALYSIS

Case Study of the SWP Phase 3 Project: Farnsworth Unit CO2-EOR Demonstration

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Introduction

To evaluate potential risks to Underground Source's of Drinking Water (USDWs) due to CO₂ leakage, a response surface methodology (RSM) based on a generic TOUGHREACT geochemical model is developed to quantify isks due to CO₂ intrusion. The case study example is the Ogallala aquifer overlying the Farnsworth unit (FWU). The main objectives of this study are to: (1) understand how CO₂ leakage is likely to influence geochemical processes in aquifer sediments; (2) quantify the potential risks to the Ogallala groundwater aquifer associated with changes in groundwater chemistry due to CO₂ leakage from the FWU; and (3) identify water chemistry factors as markers for early detection criteria.

Methodology

Simulation of cement hydration

A batch simulation of cement hydration was conducted for 100 years, to obtain the equilibrium wellbore cement composition in the field environment.

Primary chemical equations in this simulation:

- $(CaO)_3 \cdot A_2O_3 + 3CaSO_4 \cdot 2H_2O + 26H_2O \rightarrow (CaO)_6 \cdot Al_2O_3 \cdot (SO_3)_3$ $(CaO)_3 \cdot Al_2O_3 \cdot (SO_3)_3 + Al_2O_3 + Al_2O$
- $\begin{array}{l} (CaO)_{6}^{\circ} \cdot Al_{2}^{\circ} O_{3} \cdot (SO_{3})_{3}^{\circ} \cdot (H_{2}^{\circ} O)_{32} + 2(CaO)_{3}^{\circ} \cdot Al_{2}^{\circ} O_{3}^{\circ} + 22H_{2}^{\circ} O \rightarrow 3(CaO)_{4}^{\circ} \cdot Al_{2}^{\circ} O_{3}^{\circ} \cdot SO_{3}^{\circ} \cdot (H_{2}O)_{18} \\ (CaO)_{3}^{\circ} \cdot SiO_{2}^{\circ} + H_{2}O \rightarrow CSH^{\circ} + Ca(OH)_{2} \end{array}$
- $(CaO)_3 \cdot SiO_2 + H_2O \rightarrow CSH + Ca(OH)_2$ $(CaO)_2 \cdot SiO_2 + H_2O \rightarrow CSH + Ca(OH)_2$
- $(CaO)_4 \cdot Al_2 O_3 \cdot Fe_2 O_3 + 3CaSO_4 \cdot 2H_2 O + 21H_2 O \rightarrow (CaO)_6 \cdot Al_2 O_3 \cdot (SO_3)_3 \cdot (H_2 O)_{32} + Fe_2 O_3 \cdot (H_2 O)_3$

Simulation of CO_2 leakage flux through wellbores

A vertical 1-D simulation was conducted for 100 years, calibrated with the mineralogy result of the cement hydration model A dual-permeability continuum was developed to simulated fracture and matrix. The uncertainties of leakage flux were quantified by RSM.

Property	Value	Property	Value	
Rese rvo ir		USDW aquifer		
Thickness	30 m	Thickness	100 m	
Permeability	$4.52 \times 10^{-14} m^2$	Permeability	5.0×10 ⁻¹³ m ²	
Porosity	0.1453	Porosity	0.30	
Rock density	2650 kg/m ³	Rock density	2600 kg/m ³	
Salinity	0.06	Salinity	0.00035	
Pressure	300-350 bar			
CO ₂ saturation	0.1-0.9			
Cement		Fracture		
Permeability	5.0×10 ⁻²⁰ m ²	Permeability	1.0×10 ⁻¹⁴ m ²	
Porosity	0.08	Porosity	0.40	
Rock density	2400 kg/m ³	Rock density	2400 kg/m ³	
		Fracture proportion	0.01-0.1	

Quantification of the risks with CO_2 leakage into the USDW

Model setup: 2-D radial modal.	Table: Selection of the initial water chemistry			
	Name	Concentrat io n (mg/L)	Name	Concentrat io n (mg/L)
Porosity = 0.3	H ⁺	pH 7.73	Cu ²⁺ Ph ²⁺	0.014 8 7 × 10 ⁻⁴
230 grid cells	H ₂ AsO ₄ ⁻	0.0024 (as As)	HSeO3	0.003 (as Se)
CO2 Leaking	Cr(OH) ₂ *	0.002 (as Cr)	Ag ⁺ Zn ²⁺	3.61 × 10 ⁻³ 0.45
	Mn ²⁺	0.0077	TDS	570

Table: Independent parameters for simulations of CO ₂ leaking into USDW aquifer									
	Parameter name	Low (-1)	Mid (0)	High (+1)	Distribution				
	Permeability: m ²	10 ⁻¹⁴	10 ⁻¹³	10 ⁻¹²	Lognormal				
	Leaking rate: kg/(s·m ²)	10-21	10-19	10-17	Lognormal				
	CEC: meq/100g	1	6.5	12	Uniform				
	Absorbent SSA	1	50.5	100	Uniform				



Cement hydration

With the consumption of alite, belie, ferrite and aluminate, CSH, portlandite and ettringite precipitate as main products. $\begin{bmatrix}
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Figure: Composition changes during cement hydration

CO₂ leakage flux through wellbores

Results suggest a probability > 90% that any leakage would be between $10^{-14} - 10^{-10}$ kg/(m²-year). This leakage rate is far less than previous estimates. We determined that these resulting small flux values are due to: (1) reactions between CO₂ and portlandite (Ca(OH)₂) consume CO₂; (2) porosity decreases due to precipitation of CaCO₃, which further degrades the leakage pathway.



Risk assessment of CO₂ leakage into the Ogallala USDW

The coefficient of determination (\mathbb{R}^2) between simulation results and the responses of the regression (\mathbb{R} SM) model for most water indices of interest exceeded 0.9, suggesting that the resulting trained ROMs are effective for representing full-scale (non-ROM) simulation results.



Conclusions

- Probability > 90% that the leakage rate is between $10^{-14} 10^{-10}$ kg/(m²·year). Simulated small CO, leakage rate is attributed to CO, reaction with portlandite (C
- Simulated small CO₂ leakage rate is attributed to CO₂ reaction with portlandite (Ca(OH)₂) in the wellbore cement; this reaction consumes CO₂ and further decreases fracture porosity, degrading the CO₂ pathway.
- Within the range of simulated CO₂ leakage rates, trace metal concentrations could be twice as much as the initial value after 200 years, for the "worst case" scenario. However, all resulting simulated trace metal concentrations, even for the worst case scenario, are less than 1/5 of the MCL limit, suggesting that the risk of such contamination to the Ogallala groundwater quality is not significant.
- Water quality at 100, 500 and 1000 m away from the leakage pathway reflects the slight impact after 200 years, suggesting that the monitoring wells in place today will be capable of detecting significant leakage.

TDS and nitrate are suggested markers for early detection.

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Simulated water quality results

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